

## Material Safety Data Sheet

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### Section I - Chemical Product And Company Identification

#### Product Name: Sodium Hypochlorite

CAS Number: 7681-52-9

HBCC MSDS No. CC17000



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### Section II - Composition/Information On Ingredients

Chemical Name	CAS Number	%	Exposure Limits (TWAs) in Air		
			ACGIH TLV	OSHA PEL	STEL
Sodium Hypochlorite	7681-52-9	5-15	N/A	N/A	2mg/m <sup>3</sup>
Sodium Hydroxide	1330-73-2	< 1	N/A	N/A	2mg/m <sup>3</sup>

### Section III - Hazard Identification

**Routes of Exposure:** Sodium hypochlorite may affect the body either through ingestion, inhalation, or contact with the eyes and/or skin.

#### Summary of Acute Health Hazards

**Ingestion:** May cause irritation of the membranes of the mouth and throat, stomach pain, and possible ulceration.

**Inhalation:** May cause burns, cough, pulmonary edema, up to 48 hours after exposure.

**Skin:** May cause moderate skin irritation and reddening of the skin. Prolonged exposure may cause burns, blistering.

**Eyes:** May cause severe irritation such as burns, and eye damage.

**Carcinogenicity Lists:** No NTP: No IARC Monograph: No OSHA Regulated: No

**Summary of Chronic Health Hazards:** Irritating effects increase with strength of solution and time of exposure.

**Medical Conditions Generally Aggravated by Exposure:** N/A

**Note to Physician:** The absence of visible signs of burns does NOT reliably exclude the presence of actual tissue damage.

### Section IV - First Aid Measures

**Ingestion:** Do not give any liquid to an unconscious person. Drink large quantities of gelatin solution or milk. If these are not available, drink large quantities of water. Do

NOT give vinegar, baking soda or acidic antidotes. GET MEDICAL ATTENTION IMMEDIATELY.

**Inhalation:** If adverse effects occur, remove to fresh air. Give artificial respiration if not breathing. If breathing is difficult, oxygen should be administered by qualified personnel. If respiration or pulse has stopped, have a trained person administer Basic Life Support (Cardio-Pulmonary Resuscitation/ Automatic Defibrillator) and GET MEDICAL ATTENTION IMMEDIATELY.

**Skin:** Immediately flush contaminated areas with plenty of water. Remove contaminated clothing, jewelry, and shoes immediately. Wash contaminated areas with soap and water. Thoroughly clean and dry contaminated clothing and shoes before reuse. GET MEDICAL ATTENTION IMMEDIATELY.

**Eyes:** Flush with plenty of water for 15 minutes, lifting the lower and upper lids occasionally. GET MEDICAL ATTENTION IMMEDIATELY. Contact lenses should not be worn when working with this chemical.

### Section V - Fire Fighting Measures

**Flash Point:** Nonflammable

**Autoignition Temperature:** N/A

**Lower Explosive Limit:** N/A

**Upper Explosive Limit:** N/A

**Unusual Fire and Explosion Hazards:** Heat and acid contamination will produce irritating and toxic fumes. May decompose, generating irritating chlorine gas.

**Extinguishing Media:** Use extinguishing agents appropriate for surrounding fire.

**Special Firefighting Procedures:** Wear NIOSH approved positive-pressure self-contained breathing apparatus. Move container from fire area if it can be done without risk. Avoid inhalation of material or combustion by-products. Stay upwind and keep out of low areas.

### Section VI - Accidental Release Measures

**[Spills may need to be reported to the National Response Center (800/424-8802) DOT Reportable Quantity (RQ) is 100 pounds]** Ventilate the area of the spill or leak. For large spills, evacuate the hazard area of unprotected personnel. Wear appropriate protective clothing. Dike and contain. Neutralize with sodium sulfite, bisulfite or thiosulfite. Remove with vacuum trucks or pump to storage vessels. Soak up residue with an absorbent such as clay, sand or other suitable material; place in non-leaking containers for proper disposal. Flush area with water to remove trace residue; dispose of flush solutions as above. For small spills, take up with an absorbent material and place in non-leaking containers; seal tightly for proper disposal. This material is alkaline and may raise the pH of surface waters with low buffering capacity.

### Section VII - Handling and Storage

Store in vented, closed, clean non-corrosive containers in a cool, dry location away from direct sunlight and not adjacent to chemicals which may react with the bleach if spillage occurs. If closed containers become heated, the containers should be vented to release decomposition products (mainly oxygen under normal decomposition). Do not mix or contaminate with ammonia, hydrocarbons, acids, alcohols or ethers.

## Section VIII - Exposure Controls/Personal Protection

**Respiratory Protection:** Not required under normal use conditions. In the case of a fire use self-contained breathing apparatus. A NIOSH approved respirator with N95 (dust, fume, mist) filters may be permissible under certain circumstances where airborne concentrations are expected to exceed exposure limits, or when symptoms have been observed that are indicative of overexposure.

When decomposition products exist, acid gas cartridges are also required.

A half-piece air-purifying respirator may be used in concentrations up to 10X the acceptable exposure level and a full facepiece air-purifying respirator may be used in concentrations up to 50X the acceptable exposure level.

Supplied air should be used when the level is expected to above 50X the acceptable level, or when there is a potential for uncontrolled release.

A respiratory program that meets 29 CFR 1910.134 must be followed whenever workplace conditions warrant use of a respirator.

**Ventilation:** No special ventilation is required unless bleach is exposed to decomposition conditions, i.e. heat or acidic conditions.

**Protective Clothing:** Avoid contact with the eyes. Wear chemical goggles and/or face shield if there is the likelihood of contact with the eyes. Avoid prolonged or repeated contact with the skin. Wear chemical-resistant gloves and other clothing as required to minimize contact.

**Other Protective Clothing or Equipment:** Safety showers and eyewash fountains should be available in storage and handling areas.

**Work/Hygienic Practices:** All employees who handle sodium hypochlorite should wash their hands before eating, smoking, or using the toilet facilities.

## Section IX - Physical and Chemical Properties

**Physical State:** Liquid

**pH:** 12

**Freezing/Melting Point/Range:** -5 to -25°C

**Boiling Point/Range:** 40-76°C (104-169°F)  
(Decomposes)

**Appearance/Color/Odor:** Colorless to pale yellow watery liquid with a pungent chlorine odor

**Solubility in Water:** 100%

**Vapor Pressure(mmHg):** 12-17 @ 20°C

**Specific Gravity(Water=1):** 1.07-1.26 @ 20°C

**Molecular Weight:** 75.45

**Vapor Density(Air=1):** 2.61

**% Volatiles (by volume):** Variable-Water plus products of Decomposition

**How to detect this compound :**

N/A

## Section X - Stability and Reactivity

**Stability:** Unstable above 40°C, in sunlight, or in contact with acid.

**Hazardous Polymerization:** Will Not Occur

**Conditions to Avoid:** Stability decreases with concentration, heat, light exposure,

decrease in pH and contamination with heavy metals, such as nickel, cobalt, copper and iron.

**Materials to Avoid:** Strong acids, oxidizable materials, heavy metals(which act as catalysts), reducing agents, ammonia solutions, ether, and many organic and inorganic chemicals such as paint, kerosene, paint thinners, shellac, grease and oils.

**Hazardous Decomposition Products:** Chlorine. Additional decomposition products which depend upon pH, temperature and time are sodium chloride, sodium chlorate and oxygen.

### Section XI - Toxicological Information

**Toxicity Data:** By ingestion, Grade 1: oral rat LD<sub>50</sub> = 8.91 g/kg IDLH Value: Data not available

### Section XII - Ecological Information

This material may be harmful to aquatic life in low concentrations.

### Section XIII - Disposal Considerations

Can be neutralized with weak reducing agents such as sodium sulfite, bisulfite, or thiosulfite (DO NOT USE SULFATES OR BISULFATES). Dispose of in accordance with all applicable local, county, state and federal regulations.

### Section XIV - Transport Information

**DOT Proper Shipping Name:** Hypochlorite Solutions

**DOT Hazard Class/ I.D. No.:** 8, UN1791, III

### Section XV - Regulatory Information

**Reportable Quantity:** 100 Pounds (45.4 Kilograms)

**NFPA Rating:** Health - 3; Fire - 0; Reactivity - 1

0=Insignificant 1=Slight 2=Moderate 3=High 4=Extreme

**Carcinogenicity Lists:** No **NTP:** No **IARC Monograph:** No **OSHA Regulated:** No

Certified to NSF/ANSI Standard 60 12.5% Solution Maximum Use 84 mg/L

### Section XVI - Other Information

**Hazardous Ingredients:** Sodium hypochlorite is manufactured only in solution form. Industrial grade sodium hypochlorite contains from 10 - 15% by weight NaOCL (10 - 17.8% available chlorine) with about 0.50-1.00% excess NaOH for stability control.

**Synonyms/Common Names:** Liquid Bleach

**Chemical Family/Type:** Halogen Compound

**Sections changed since last revision:** XV

**IMPORTANT!** Read this MSDS before use or disposal of this product. Pass along the information to employees and any other persons who could be exposed to the product to be sure that they are aware of the information before use or other exposure. This MSDS has been prepared according to the OSHA Hazard Communication Standard [29 CFR 1910.1200]. The MSDS information is based on sources believed to be reliable.

However, since data, safety standards, and government regulations are subject to change and the conditions of handling and use, or misuse are beyond our control, [Hill Brothers](#)

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